Balancing Chemical Equations <u>Balancing Chemical Equations Simulation</u> Author: Jackie Esler

Balancing a Chemical Equation Pre-lab

Name:

pre-lab

1. Aria wanted to design a new bouncy-chew toy for his new puppy. She used this "equation" to build it. The arrow  $\implies$  shows what she had "before" and "after" building her new toys.

	$\infty$	$\rightarrow$	
a) Did Aria balance her equation?	Yes	no	not sure

b) Not so fast ... Explain your answer to number 1!

I chose \_\_\_\_\_\_ because \_\_\_\_\_

- c) The "before" objects are called **reactants**. Go back to the picture of the equation and circle all of the reactants , put (R) in the circle(s).
- d) The "after" objects are called products. Go back to the picture of the equation and box around all products.
  Put a (P) in the box(es).
- **2.** In the following chemical equation:

 $CO_2 + H_2O \rightarrow C_2H_6 + O_2$ 

b) Is this chemical equation balanced?

a) List the products: \_\_\_\_\_

- > Yes, because \_\_\_\_\_
- No, because \_\_\_\_\_\_