

Balancing	g Ch	nemic	al Equa	itions		Stude	ent Guide	<b>:</b>		Name: _		
Start:	1. 2. 3. 4.	Good Click Click Ty bot a. T	ogle: F k on the c on the	PhET Bare first li	ink  Now! the sire  The Si	Chemicoutton. mulation	cal Equa	tions		on everyth		
Tell mo										is the purp		se tools? 
0- N <sub>2</sub>		+		0-1	H <sub>2</sub>				0	₽ NH <sub>2</sub>		
6.	Is	this e	:quatic	n balar	nced? {	Explain	ı					
Change t					•	on unt	il you ge	et a 😃	ļ			
											eacher che	ck-off

1

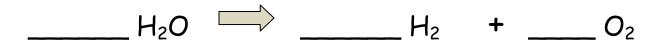
Balancing Chemical Equations using the <u>Balancing Chemical Equations Simulation</u> Author: Jackie Esler





7. Is this equation balanced? Explain.

Using the simulation- add coefficients so that the equation is balanced correctly.



Using the simulation- add coefficients so the equation is balanced correctly.

$$CH_4 + O_2 \longrightarrow CO_2 + H_2O$$
List all reactants:

\_\_\_\_\_

9. Discuss the maximum (highest number) of ammonia molecules the simulation will let you make? Draw them:

In the product, ammonia ( $2NH_3$ ), What does the (2) tell you about the equation?

What does the (3) tell you?

Teacher check-off/check-in... Go on to Balancing Game tab...